National Aeronautics and **Space Administration** 



# Sabatier Subsystem Thermal Management

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## Sabatier Subsystem Overview



- **Basic system consist of the Sabatier reactor, condenser, and thermal control system**
- **Reactor is used to convert carbon dioxide and hydrogen into methane and water**
	- $CO<sub>2</sub> + 4H<sub>2</sub> \rightarrow CH<sub>4</sub> + 2H<sub>2</sub>O$
	- **Exothermic, ΔH = -165.4 kJ/mole**
- **Condenser is used to separate the methane and water**
- **Single or multiple Reactors, each can be Adiabatic or Isothermal**
	- Single reactor can have both adiabatic and isothermal zones or a thermal gradient along length
- **Adiabatic Reactor** 
	- Operates at higher temperature for faster kinetics (smaller reactor)
	- Recycled product gas can be used to control reactor temperature
- **Isothermal Reactor**
	- Operates at lower temperatures for higher CO2 conversion efficiency
	- Thermal control system required to removed heat from reactor



#### **Advanced Designs**







# Basic Thermal Control System (TCS)



- Cold reactant feed is typically preheated directly by reactor
- Isothermal reactor operates at ~400°C, adiabatic reactor temperature must kept < 600°C
- Reactor coolant loop removes up to 50% of total heat load
- Condenser temperature is maintained just above  $H<sub>2</sub>O$  freezing point
- Condenser coolant loop removes up to 100% of total heat load (adiabatic reactor)
	- Radiator at reactor exit would likely be used to help cool the reactor products before entering the condenser

# Thermal Challenges



- **Reactor and condenser operate at significantly different temperatures**
	- The reactor at around 400 °C and the condenser at ~5°C
	- A single coolant loop is desired to minimize TCS mass and increase reliability
- **High interdependence between reactor and thermal control system designs**
	- Reactor temperature has a significantly impact on performance
	- Certain reactors have unique requirements that have advantages or disadvantages that do not become apparent until considered as a system
		- Example: Is it better to drive the Sabatier reactor to high conversion with thermal management or use gas recycling and lower conversion with no thermal management?
- **Reactor is capable of generating a significant amount of heat per unit area** 
	- TCS must minimize risk of hot spots (catalyst deactivation) or reaction quenching
- **Continuous Operation**
	- TCS must adjust to fluctuations in feed gas temperature and ambient temperature
	- Large ambient temperature range -112 to 5°C
- **Condenser must produce liquid water near freezing temperature** 
	- 2-5°C desired to minimize partial pressure of water vapor in product gases
	- Low temperature delta between condenser and environment for several hours per day, active cooling will likely be required to reject heat during the hottest part of the day
- **Low passive heat transfer due to low atmospheric pressure and gravity**

### Potential Solutions – System Level



# Potential Solutions: Isothermal Reactor



- **Main Features:** High heat exchange efficiency (i.e. reduce weight and size) and controllability (e.g. self-sustainability) of the catalyst temperature.
- **Possible Reactor Designs:** Packed bed (traditional), microchannel, monolith, microlith, etc.
- **Reactor Heat Exchanger Designs:** 
	- Single cooling jacket (shell)
	- Multiple cooling channels (tubes)
	- Cocurrent or countercurrent flow
- **Reactor Temperature Profile:** 
	- Uniform: requires liquid coolant boiling at high pressure or high coolant flow rate
	- Variable: more difficult to control
- **Heat Exchange Fluid:**
	- Gas: CO2, Hydrogen, Methane, etc.
		- high flow rates required (larger HEX)
	- Liquid: Water, Dowtherm, Oil, Molten Salt, etc.
		- Lower flow rate, isothermal boiling possible
- **Heat can be rejected to ambient by flowing coolant thru a radiator**
	- Refrigeration cycle is not required due to high temperature delta
	- Direct air cooling may not be possible



# Potential Solutions: Condenser

• **Main Features:** High heat exchange efficiency (i.e. reduce weight and size) and controllability. Minimal power input.

#### **Cooling Options**

- **Direct air cooling or Liquid (e.g. antifreeze) cooling loop** 
	- Pros: Very low energy input, high simplicity/reliability
	- Cons: Very low control, likely not feasible due to low dT between max ambient & condenser target temperature
- **Thermoelectric cooler (TEC)**
	- Pros: High simplicity/reliability, high temperature control
	- Cons: Poor power efficiency (6x higher power input relative to conventional vapor-compression), Low temperature delta (large radiator or multistage coolers required)
- **Vapor-Compression cycle** 
	- Pros: Good efficiency and temperature control. Saturated CO2 can be used isothermally cool condenser. Same system could also be used to cool an isothermal reactor.
	- Cons: Less reliable than TEC. Will require higher pressure condenser and radiator.
- **Vapor-Absorption cycle** 
	- Pros: Low electrical power input, utilizes waste heat from reactor and other systems
	- Cons: Complex, ammonia refrigerant required. May not be feasible.





## Potential Solutions: Condenser

#### **Heat Exchanger Design Options**

- **Plate HEX** 
	- Thin metal plates used to transfer heat between fluids
	- High heat transfer efficiency, small size
	- Higher pressure drop than shell-and-tube
- **Shell and Tube** 
	- Most commonly used heat exchanger
	- Suited for high-pressure applications
	- Lower efficiency than plate heat exchangers
- **Submerged Coil**
	- Products cooled by flowing through a coil submerged in a stagnant heat transfer liquid; similar to shell and tube except heat is extracted at the condenser outer wall
	- This design is many applicable to thermoelectric cooling
- **Direct Contact (low potential)**
	- Reactor gases cooled by mixing with coolant. Coolant must be the same as product (CH4) to avoid contamination
	- Very high heat transfer, simplest condenser design
	- May not be feasible due to high methane flow rate. Very cold coolant required, high risk of H2O freezing.

Reactor **Products NCG** Coolant 0°C← **H20** 







### Backup