National Aeronautics and Space Administration



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Determining Thermal Management Requirements for an ISRU Plant using a Reduced Order Model

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System thermal model progression:

1) Single point – steady state

- Easy to build in excel.
- Good for mission trades.
- Easy to swap components.

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3) High fidelity – steady state / transient

- Geometrically represent many components.
- Use Thermal Desktop or similar software.
- Predict worst case temperatures.
- Flesh out the system design.

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2) Reduced order - Transient

- Identify thermal requirements.
- Gives higher fidelity predictions.

• Needed to construct Step 3.

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System thermal model progression:

Current point of interest

1) Single point – steady state

- Easy to build in excel.
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System thermal model progression:

1) Single point – steady state

- Easy to build in excel.
- Good for mission trades.
- Easy to swap components.

Not a lot of variability

3) High fidelity – steady state / transient

- Geometrically represent many components.
- Use Thermal Desktop or similar software.
- Predict worst case temperatures.
- Flesh out the system design.

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System thermal model progression:

Several different options for obtaining this information.

- ASPEN Custom Modeler
- MATLAB/SIMULINK
- SINDA/FLUINT

2) Reduced order - Transient

- Identify thermal requirements.
- Gives higher fidelity predictions.
- Needed to construct Step 3.



Propose: Use Cantera Python module.

- Freely available, open source, extensible.
 - Originated from Caltech.
- Designed for thermodynamics, transport processes, chemical kinetics problems.^[1]

(i) Not secure | www.cantera.org/docs/sphinx/html/about.html



David G. Goodwin, Harry K. Moffat, and Raymond L. Speth. *Cantera: An object-oriented software toolkit for chemical kinetics, thermodynamics, and transport processes*. <u>http://www.cantera.org</u>, 2017. Version 2.3.0. doi:10.5281/zenodo.170284



Plant Overview:

- ISRU plant on a predeployed lander produces fuel prior to crew arrival.^[2]
 - Option 1: LOX (~435 days)
 - Option 2: LOX and LCH₄ (~435 days)
- 3 independent modules.
- If one failed, 2 still produce enough propellant.^[2]





Evolvable Mars Campaign (EMC) Mars Ascent Module (MAV) concept.^[3]

Notional ISRU system packaging – produced for EMC.^[4]



Objective:

- Ensure components stay within their temperature limits.
- Minimize required power and radiator size.
 - E.g. Leverage environment, waste heat, energy storage.



Evolvable Mars Campaign (EMC) Mars Ascent Module (MAV) concept.^[3]

Notional ISRU system packaging – produced for EMC.^[4]



Modeling Challenges:

- More dynamic than most chemical plants.
 - Mars day/night temperature swing.
 - Could vary production rate.
- Steady state models can over predict heating/cooling needs.



Mars sky temperature (left) and inlet solar flux (right) for a location in Jezero crater during summer solstice with dust optical depth of 0.3. Interpolated form data provided by ref [5].



Modeling Challenges:

- High power and heat rejection for a payload.
- Need to identify heat recuperation options.

Results from NASA EMC's steady state excel models of two plant types^[4]

435 day comparison Option 1 (LOX only):

- 14kW heat rejection.
- Uses a CO₂ freezer and solid oxide electrolyzer.

Option 2 (LOX + LCH_4):

- 17.5kW heat rejection
- Uses a CO₂ freezer, sabatier reactor and water electrolyzer.



Output from EMC model for 435 production days – provided by Paz .^[4] Waste heat = sustained worst case heat rejection (300K environment).

ISRU Plant Modeling Challenges



Modeling Challenges:

- Lots of design variability:
 - LOX or LOX + LCH₄.
 - Various possible technologies (e.g. SOE, RWGS, Sabatier).



Block diagram for ISRU plant LOX production with an Solid Oxide Electrolyzer (SOE).^[6]

ISRU Plant Modeling Challenges



Modeling Challenges:

- Lots of design variability:
 - LOX or LOX + LCH₄.
 - Various possible technologies (e.g. SOE, RWGS, Sabatier).



Block diagram for ISRU plant LOX production with an Reverse Water Gas Shift Reactor (RWGS).^[6]

ISRU Plant Modeling Challenges



Modeling Challenges:

- Lots of design variability:
 - LOX or LOX + LCH₄.
 - Various possible technologies (e.g. SOE, RWGS, Sabatier).



Block diagram for ISRU plant LOX & LCH₄ production with an Sabatier reactor.^[6]

Cantera Process Flow



Modeled component:	Cantera Object:			
Lumped mass	ReactorContains: solid, liquid, or gasCan specify ideal gas or constant pressure			
Boundary	ReservoirSpecify: contents, temperature, pressureCan dynamically update			
Thermal –⁄///– Conductance	 Wall Specify: area, conductance, object emissivity for simple enclosure problems, additional heat load 			
Mass flow	Mass Flow Controller • Specify flow rate (must be ≥ 0) Pressure controller (pair with Mass Flow Controller) • Specify valve coefficient [kg/(s Pa)] • $\dot{m} = \dot{m}_{set upstream} + K_v(P_1 - P_2)$ Valve • Specify valve coefficient [kg/(s Pa)] • $\dot{m} = K_v(P_1 - P_2)$			

Cantera Process Flow



Example Cantera Model Structure:



Heat Transfer Network

Thermal Desktop Model

Heat Load on Tank

Mass in Tank





Reactor energy conservation: $\frac{dU}{dt} = -p\frac{dV}{dt} - \dot{Q}_{net walls} + \sum_{n=i}^{N} \dot{m}_{in}h_{in} - h\sum_{n=i}^{N} \dot{m}_{out}$

Wall heat transferred between reactors: $\dot{Q}_{wall} = UA(T_{left} - T_{right}) + \varepsilon \sigma A(T_{left}^4 - T_{right}^4) + Aq_{fn}(t)$

Cantera Model vs. Thermal Desktop:

- Heat water in a 36[in] long steel tube.
 - Outer Diameter: 0.5[in], Inner Diameter: 0.3 [in].
- Cantera model:
 - Water constant pressure reactors.
 - Steel steel filled reactors.
 - Doesn't calculate pressure drop from friction.
 - Convection coefficient, G_{conv}, is an input (used correlations in SINDA v6 manual Appendix B1).



Thermal Desktop Model.



Heat Transfer Network.



Cantera Model vs. Thermal Desktop:

• Heat water in a 36[in] long steel tube.

	Tf0	Ts0	Tf1	Ts1	Tf2	Ts2	Tf3	Ts3
Cantera	300.85	306.81	301.71	307.61	302.56	308.42	303.42	309.22
TD	300.85	306.81	301.71	307.61	302.56	308.42	303.42	309.22
delta	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00

Node >320 +1/Custom View1/THERMAL PP1 4 Lump >303.4 320 303.4 318.7 300. 🤹 303.1 317.4 302.7 316 301. 7 302.4 314.7 .307.6 302.1 313.4 302. 6 301.7 312.1 .308.4 301.4 310.8 303. 4 301 309.4 _309. 2 300.7 308.1 300.3 306.8 300 < 306. Temperature [K], Time = 30 sec < 300 Temperature [K], Time = 30 sec

Fluid properties for Gconv, were calculated at fluid temperature (easy point of comparison).



Heat up comparison.

Thermal Desktop Model.



Cantera Model vs. Thermal Desktop:

Because heat loads are applied at walls Q_{TD} does not always equal Q_{wall}.

How to determine Q_{wall}?

Write out 2 energy balances:

- 1. Apply heat loads, Q_{TDs,} directly to nodes.
- 2. Add heat loads, Q_{walls} , to walls between nodes (Cantera wall governing equation). Find what Q_{wall} to Q_{TD} relationship solves to the same equation.



Temperature [K], Time = 18000 sec

Thermal Desktop Model.



Heat Transfer Network.

- Q_{wall31}=Q_{TD1}
- Q_{wall42}=Q_{TD2}
- $Q_{wall53} = Q_{TD1} + Q_{TD3}$
- $Q_{wall54} = Q_{TD2} + Q_{TD4}$
- $Q_{wallb3} = Q_{TD5} + Q_{TD4} + Q_{TD3} + Q_{TD2} + Q_{TD1}$

 \mathbf{Q}_{TD} to \mathbf{Q}_{wall} relationship. Valid if simple enclosure radiation assumption holds.



Cantera Model vs. Thermal Desktop:



Thermal Desktop Model Output vs Cantera Model Output

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Cantera Summary:

- Relies more on user input than Thermal Desktop SINDA/FLUINT.
 - E.g. user calculates Tsinks and conductances.
- Provides a good level of freedom to model flow of a chemical plant.
 - E.g. can set reservoir states based on user defined functions.
- Useful object oriented structure can be leveraged.
 - Can create/modify reactors in loops.
 - Can choose which components to model.
- Has built-in equations of state for liquid/water CH_4 , O_2 , and H_2O .
 - Useful for condensation/liquefaction.
- Uses thermodynamic models (NASA9, constant specific heat) to obtain enthalpy, entropy, and specific heat at a given state.
- Can obtain transport properties from:
 - CoolProp Python module
 - Transport data file^[1,7]
- Can calculate reactor products (use gri30.xml or custom reaction specifications).

Currently developing Cantera models for ISRU case: LOX production with a CO₂ freezer and Solid Oxide Electrolyzer.



Backup

Plant Level Thermal Control System



Approximate Mars Atmosphere for ISRU Ops:

- Day Temperatures: -30C to 5C
- Night Temperatures: -90 to -60C
- Pressure: 730 925 Pa
- Max wind: 15 m/s^[8,9]

From ISRU capabilities study.^[10] Actuals are landing site dependent.



Mars sky temperature and inlet solar flux for a location in Jezero crater during summer solstice with dust optical depth of 0.3. Interpolated form data provided by ref [4].



Couple reactor to Mars Environment:

- Read file with solar flux, sky, ground, and air temperatures vs. time.
- Calculate sink temperature, Tsink.
- Connect reactor to reservoir set to Tsink via a wall with radiation.
- Connect reactor to reservoir set to Tatm via will with convection.
- Update reservoirs with time.





Sample model input for coupling object to the Mars environment (temperatures updated once per hour).

Notional Lander TCS with ISRU Plant

- NASA's Evolvable Mars Campaign's 2017 concept assumed a joined thermal control system for the Mars Descent Module (MDM) and MAV.
 - Cryocoolers for ISRU fuel production/liquefaction and the propellant tank's Broad Area Cooling network both connect to a loop heat pipe system that rejects heat via radiators.



Evolvable Mars Campaign (EMC) MAV thermal control system concept. ^[3]



Cantera Model vs. Thermal Desktop:

- Heat ideal CO₂ gas in Inconel tube
- Inconel Tube
 - ε=0.69, cp= 412.41 [J/(kg K)]
 - length=36[in], OD=0.5[in], ID=0.3[in]
 - 36 divisions in tube
 - Heat per node: Q_{TD}=Q_{wall}=-100[W]
 - Expect slight in results because Thermal Desktop's pipe uses edge nodes.

=====Summary of Cantera Results at end time=====
time end (seconds) = 4000
fluid reactor 0 T [K]= 335.233617045 P [Pa]= 101325.0
solid reactor 0 T [K]= 761.64042079
tube to fluid wall 0 Qnet [W]= 12.9616379322
fluid reactor 1 T [K]= 363.188495391 P [Pa]= 101325.0
solid reactor 1 T [K]= 767.028258477
tube to fluid wall 1 Qnet [W]= 12.3729154791
fluid reactor 2 T [K]= 389.164945007 P [Pa]= 101325.0
solid reactor 2 T [K]= 773.859582912
tube to fluid wall 2 Qnet [W]= 11.811247238
fluid reactor 3 T [K]= 413.376938604 P [Pa]= 101325.0
solid reactor 3 T [K]= 780.879676064
solid reactor 3 T [K]= 780.879676064 fluid reactor 33 T [K]= 753.365353267 P [Pa]= 101325.0
solid reactor 3 T [K]= 780.879676064 fluid reactor 33 T [K]= 753.365353267 P [Pa]= 101325.0 solid reactor 33 T [K]= 872.793258829
solid reactor 3 T [K]= 780.879676064 fluid reactor 33 T [K]= 753.365353267 P [Pa]= 101325.0 solid reactor 33 T [K]= 872.793258829 tube to fluid wall 33 Qnet [W]= 3.00385484918
solid reactor 3 T [K]= 780.879676064 fluid reactor 33 T [K]= 753.365353267 P [Pa]= 101325.0 solid reactor 33 T [K]= 872.793258829 tube to fluid wall 33 Qnet [W]= 3.00385484918 fluid reactor 34 T [K]= 758.436195215 P [Pa]= 101325.0
<pre>solid reactor 3 T [K]= 780.879676064 fluid reactor 33 T [K]= 753.365353267 P [Pa]= 101325.0 solid reactor 33 T [K]= 872.793258829 tube to fluid wall 33 Qnet [W]= 3.00385484918 fluid reactor 34 T [K]= 758.436195215 P [Pa]= 101325.0 solid reactor 34 T [K]= 873.808685244</pre>
<pre>solid reactor 3 T [K]= 780.879676064 fluid reactor 33 T [K]= 753.365353267 P [Pa]= 101325.0 solid reactor 33 T [K]= 872.793258829 tube to fluid wall 33 Qnet [W]= 3.00385484918 fluid reactor 34 T [K]= 758.436195215 P [Pa]= 101325.0 solid reactor 34 T [K]= 873.808685244 tube to fluid wall 34 Qnet [W]= 2.88865847529</pre>
<pre>solid reactor 3 T [K]= 780.879676064 fluid reactor 33 T [K]= 753.365353267 P [Pa]= 101325.0 solid reactor 33 T [K]= 872.793258829 tube to fluid wall 33 Qnet [W]= 3.00385484918 fluid reactor 34 T [K]= 758.436195215 P [Pa]= 101325.0 solid reactor 34 T [K]= 873.808685244 tube to fluid wall 34 Qnet [W]= 2.88865847529 fluid reactor 35 T [K]= 763.297335443 P [Pa]= 101325.0</pre>
<pre>solid reactor 3 T [K]= 780.879676064 fluid reactor 33 T [K]= 753.365353267 P [Pa]= 101325.0 solid reactor 33 T [K]= 872.793258829 tube to fluid wall 33 Qnet [W]= 3.00385484918 fluid reactor 34 T [K]= 758.436195215 P [Pa]= 101325.0 solid reactor 34 T [K]= 873.808685244 tube to fluid wall 34 Qnet [W]= 2.88865847529 fluid reactor 35 T [K]= 874.587614997</pre>
<pre>solid reactor 3 T [K]= 780.879676064 fluid reactor 33 T [K]= 753.365353267 P [Pa]= 101325.0 solid reactor 33 T [K]= 872.793258829 tube to fluid wall 33 Qnet [W]= 3.00385484918 fluid reactor 34 T [K]= 758.436195215 P [Pa]= 101325.0 solid reactor 34 T [K]= 873.808685244 tube to fluid wall 34 Qnet [W]= 2.88865847529 fluid reactor 35 T [K]= 763.297335443 P [Pa]= 101325.0 solid reactor 35 T [K]= 874.587614997 tube to fluid wall 35 Qnet [W]= 2.77419790941</pre>

Cantera Results, Ideal GasReactor start and end of pipe – transient run to steady state.

- Fully developed flow
 - m_f= 0.000495 [kg/s]
 - T_{in}=T_{initial}= 305[K], P_{in}=101325[Pa]
 - Properties found with fluid temperature
 - Environment: ε=1, T_{env}=923.15[K]



Temperature [K], Time = O sec

Thermal Desktop Steady State Results – modeled a pipe with 36 divisions.

Ideal gas property file with temperature dependent viscosity, thermal conductivity, and constant pressure specific heat at 101325Pa.

MODR=0 (use fluid properties to calculate Nusselt number).

Sources



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Sources



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